

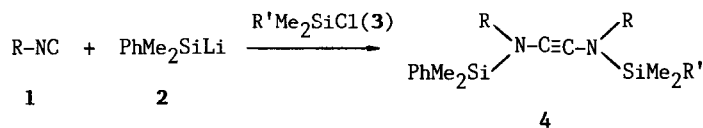
A NOVEL DIMERIZATION OF ISONITRILE BY ORGANOSILYLLITHIUM

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**Summary:** Reactions of alkyl isocyanides with phenyldimethylsilyllithium followed by treatment with trialkylchlorosilane afforded 1-[*N*-(phenyldimethylsilyl)-*N*-alkylamino]-2-[*N*-(trialkylsilyl)-*N*-alkylamino]acetylenes in moderate yields.

Synthetically useful methods have been developed on the basis of the selective lithiation of isocyanides.<sup>1)</sup> Primary- and secondary-alkyl isocyanides are selectively lithiated at the  $\alpha$  carbon atom to the isocyano group to generate  $\alpha$ -lithioalkyl isocyanides.<sup>1)</sup> On the other hand, tert-alkyl isocyanides having no  $\alpha$ -hydrogen are susceptible to  $\alpha, \alpha$ -addition with alkyllithium to give lithium aldolimine intermediates.<sup>2)</sup>

Herein, we wish to describe that alkyl isocyanides (**1**) are readily dimerized by organosilyllithium (**2**) to produce di-[*N*-(organosilyl)-*N*-alkylamino]acetylenes (**4**) in moderate yields, after quenching with trialkylchlorosilanes (**3**). The diaminoacetylene structure of **4** was unambiguously supported by Raman spectral band at  $\nu_{C\equiv C}$  2210  $\text{cm}^{-1}$ .



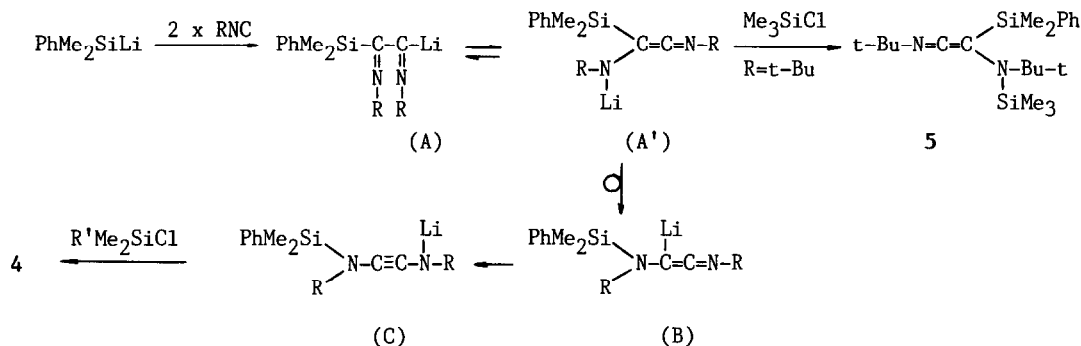
- a: R=c-C<sub>6</sub>H<sub>11</sub>, R'=Me 79%
- b: R=sec-C<sub>4</sub>H<sub>9</sub>, R'=Me 67%
- c: R=sec-C<sub>4</sub>H<sub>9</sub>, R'=Ph 70%
- d: R=iso-C<sub>3</sub>H<sub>7</sub>, R'=Me 56%
- e: R=iso-C<sub>3</sub>H<sub>7</sub>, R'=Ph 53%

A general procedure for the dimerization of isocyanide is as follows. To a stirred solution of phenyldimethylsilyllithium (3 mmol) in tetrahydrofuran (3 ml) at -20°C under nitrogen atmosphere, sec-butyl isocyanide (3 mmol) was dropwise added and stirred for 4 hrs. The reaction mixture was treated with phenyldimethylchlorosilane (4.5 mmol) and stirred for 2 hrs at room temperature, and then distilled in vacuo to afford 1,2-bis[*N*-(phenyldimethylsilyl)-*N*-sec-butylamino]acetylene (**4c**): <sup>1</sup>H NMR (CCl<sub>4</sub> with c-C<sub>6</sub>H<sub>12</sub> internal reference)  $\delta$  0.37 (s, 12H), 0.76 (t, 6H), 0.98 (d, 6H), 1.13-1.75 (m, 4H), 2.40-2.82 (m, 2H), 7.14-7.32 (broad m, 6H), 7.38-7.62 (broad m, 4H); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>)  $\delta$  -1.86 and -1.80, 11.70 and 11.72, 20.93 and 20.95, 28.53 and 28.57, 54.25 and 54.26, 66.42, 129.57, 134.28, 138.57; Exact Mass (m/e) Calcd for C<sub>26</sub>H<sub>40</sub>N<sub>2</sub>Si<sub>2</sub> 436.2728. Found 436.2722.

A similar treatment of tert-butyl isocyanide with phenyldimethylsilyllithium followed by quenching with trimethylchlorosilane afforded a desired diaminoacetylene (**4f**)<sup>3)</sup> only in 11% yield together with a keteneimine derivative (**5**)(7%), which was tentatively assigned on the basis of IR and <sup>13</sup>CNMR.<sup>4)</sup>

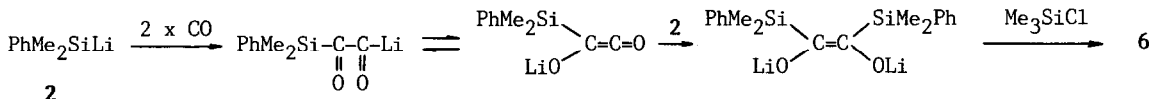
The present dimerization may be explained by a reaction scheme involving a successive insertion of two isonitriles into the lithium-silicon bond of organosilyllithium, and an anionic 1,2-migration of silyl group in the resulting intermediate (A').

Scheme 1.



Carbon monoxide, which is isoelectronic to isonitrile, also reacts with phenyldimethylsilyllithium followed by treatment with trimethylchlorosilane to afford a dimeric 1,2-bis(phenyldimethylsilyl)-1,2-bis(trimethylsilyloxy)ethylene (**6**) in moderate yield.<sup>5)</sup> Based upon the findings of the isonitrile dimerization mentioned above, the formation of **6** may be rationalized by Scheme 2.

Scheme 2.



#### REFERENCES AND NOTES

- (1) (a) U. Schöllkopf and F. Gerhart, *Angew. Chem.*, **80**, 842 (1968). (b) U. Schöllkopf, *Angew. Chem., Int. Ed. Engl.*, **16**, 339 (1977).
- (2) (a) H.M. Walborsky and G.E. Niznik, *J. Am. Chem. Soc.*, **91**, 7778 (1969). (b) Y. Ito, H. Imai, T. Matsuura, and T. Saegusa, *Tetrahedron Lett.*, **25**, 3091 (1984).
- (3) **4f**: <sup>1</sup>H NMR (CCl<sub>4</sub> with *c*-C<sub>6</sub>H<sub>12</sub> internal reference) δ 0.18 (s, 9H), 0.44 (s, 6H), 1.13 (s, 9H), 1.22 (s, 9H), 7.23-7.38 (broad m, 3H), 7.52-7.66 (broad m, 2H); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>) δ 1.9, 3.1, 30.1, 30.3, 56.3, 56.7, 70.1, 70.2, 129.4, 133.8, 141.2; Exact Mass (m/e) Calcd for C<sub>21</sub>H<sub>38</sub>N<sub>2</sub>Si<sub>2</sub> 374.2571. Found 374.2565.
- (4) **5**: IR (neat) 1965 cm<sup>-1</sup>; <sup>1</sup>H NMR (CCl<sub>4</sub>) δ 0.12 (s, 9H), 0.36 and 0.42 (2s, 6H), 1.08 (s, 9H), 1.31 (s, 9H), 7.29-7.34 (broad m, 3H), 7.43-7.72 (broad m, 2H); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>) δ -0.27, 3.07 and 3.24 and 3.25, 30.58 and 31.72, 55.76 and 57.78, 70.04, 127.85, 128.81, 134.19, 143.53, 183.93; Exact Mass (m/e) Calcd for C<sub>21</sub>H<sub>38</sub>N<sub>2</sub>Si<sub>2</sub> 374.2571, Found 374.2571. The spectral data do not exclude another possible structure; *t*-C<sub>4</sub>H<sub>9</sub>-N=C=C[N(*t*-C<sub>4</sub>H<sub>9</sub>)(SiMe<sub>2</sub>Ph)](SiMe<sub>3</sub>).
- (5) P. Jutzi and F-W. Schröder, *J. Organometal. Chem.*, **24**, C43 (1970).

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